

Chapter 2 Review

Definitions

A testable statement

A testable statement

hypothesis

$$y = 1/x$$

$$y = 1/x$$

Inverse Proportion

Generalization that
explains a body of
known facts or
phenomena

Generalization that explains a body
of known facts or phenomena

Theory

Measurement near the
accepted value

Measurement near the
accepted value

Accuracy

Repeated close
measurements

Repeated close measurements

Precision

$$y = x$$

$$y = x$$

Direct Proportion

Measure of Earth's
gravitational pull on
matter

Measure of Earth's gravitational pull on matter

Weight

Identify

Three base metric
units

Three base metric units

meter, liter, gram

The unit cm^3 measures
what quantity?

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what quantity?

Volume

Which observation is qualitative?

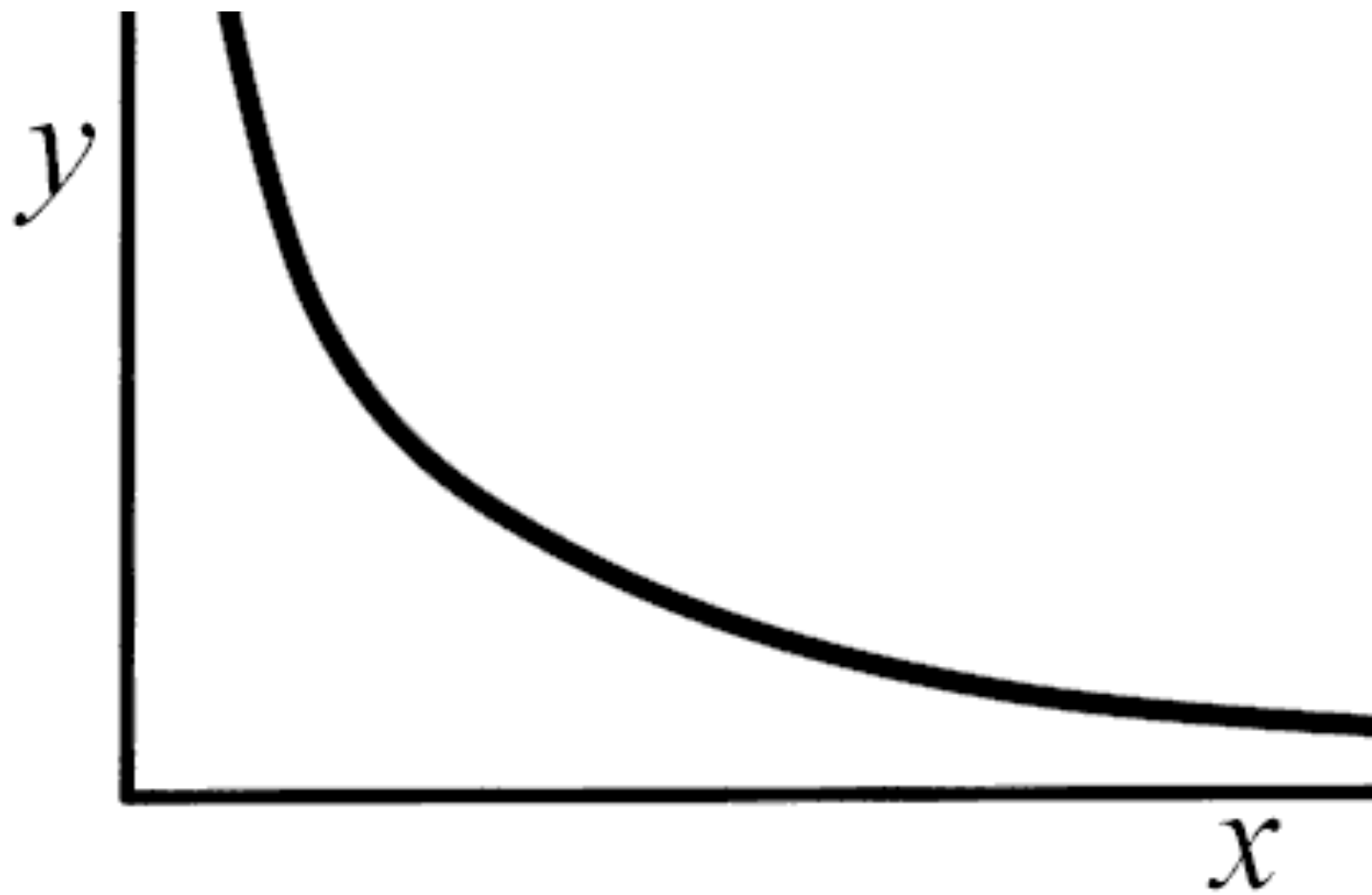
- A) The mass is 74.3 kg.
- B) The mixture is saturated.
- C) The pH of the liquid is 7.

Which observation is qualitative?

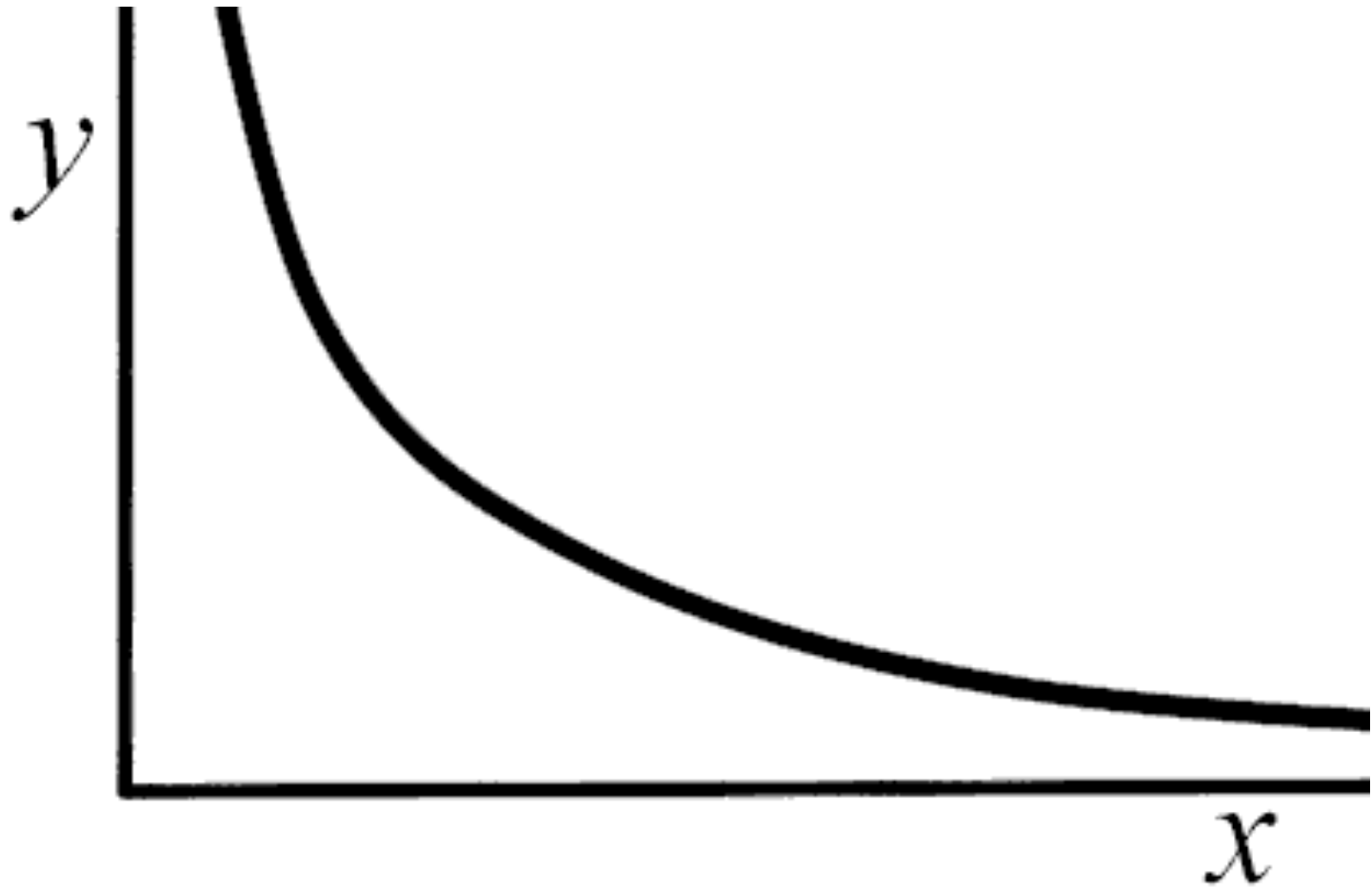
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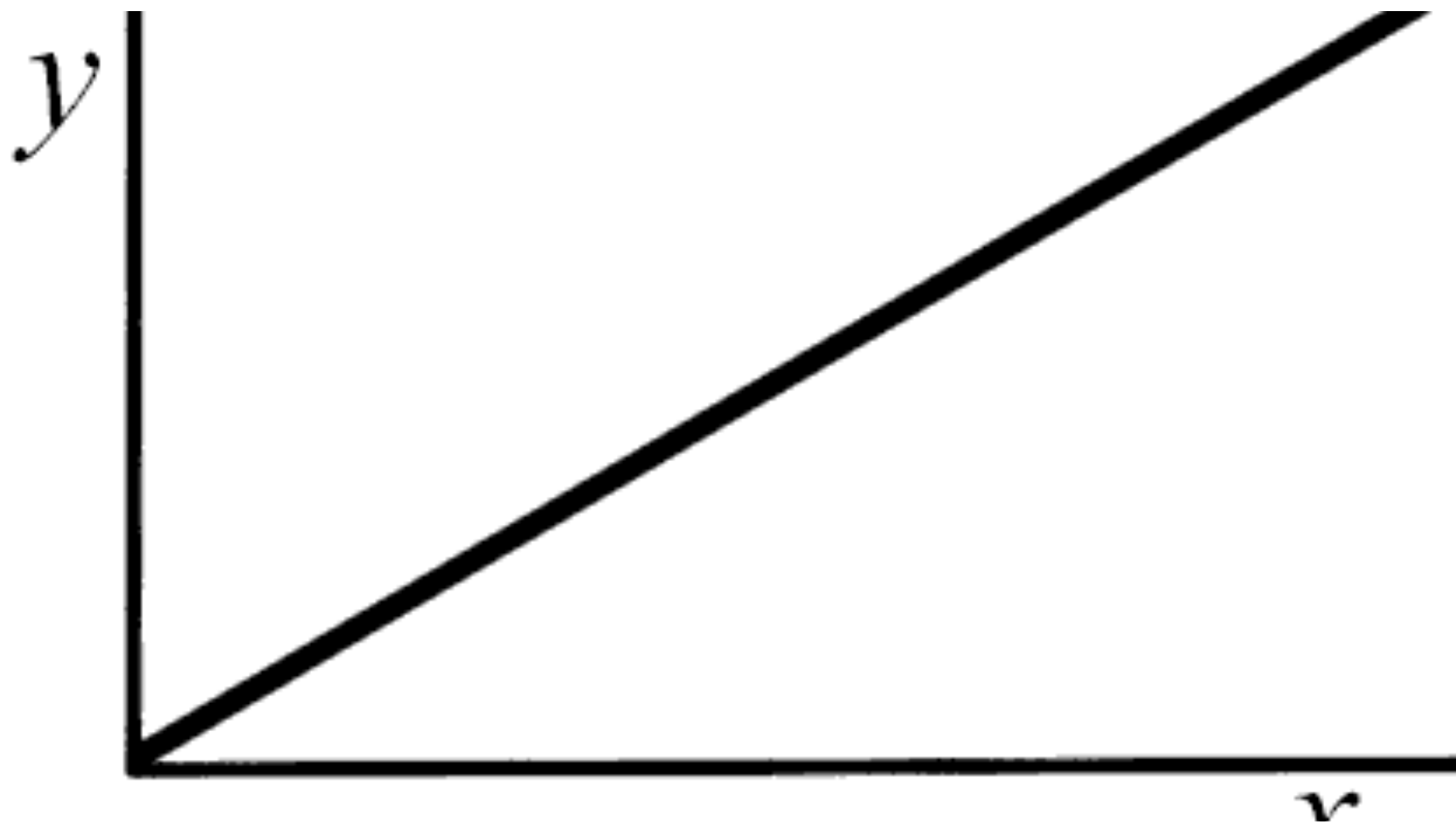
Inverse Proportion



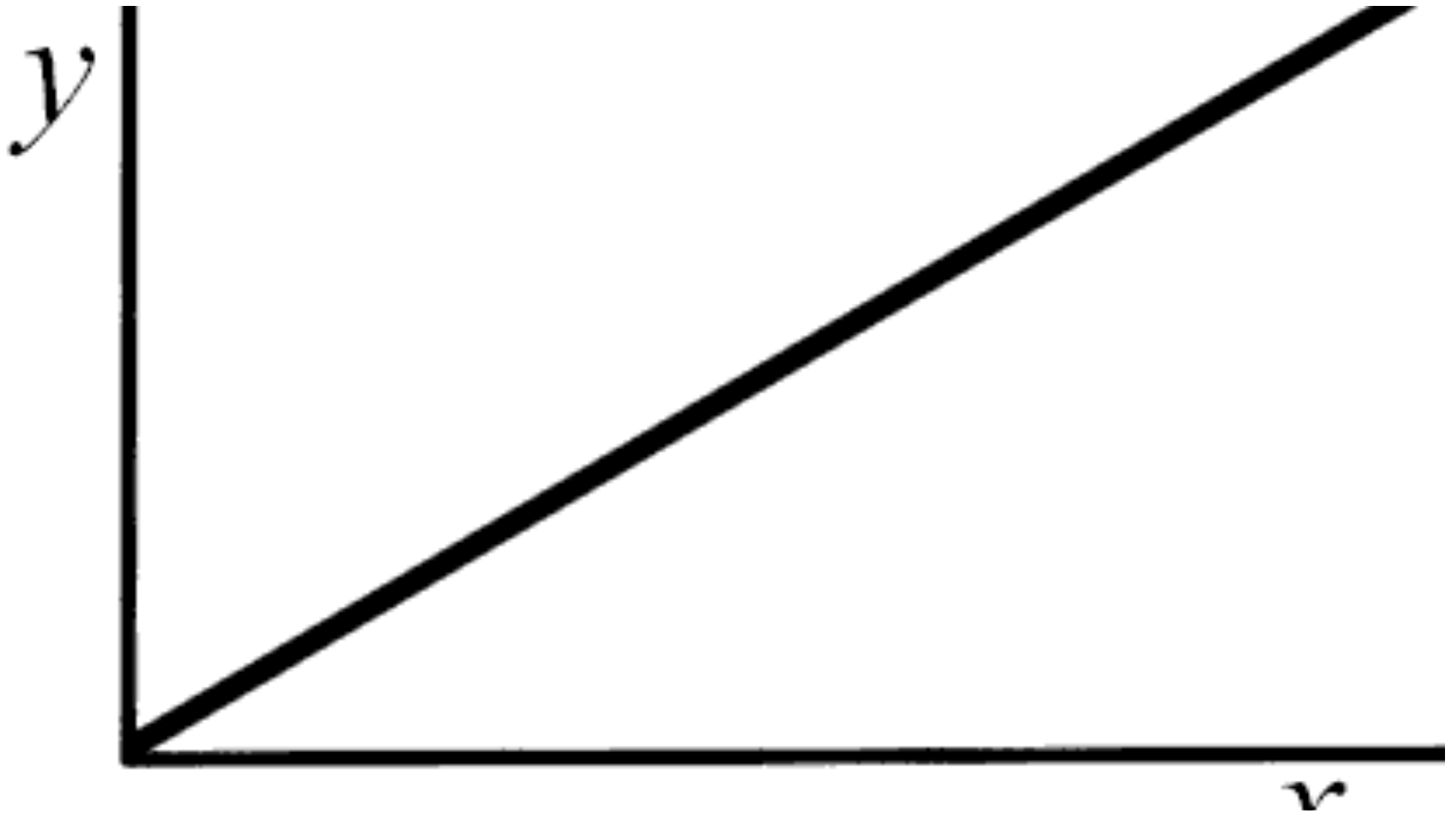
Metric unit to measure the
length of a shoe

Metric unit to measure the
length of a shoe

cm



Direct Proportion



Which factor is important to specify when measuring density?

A) Temperature

B) Length

C) Weight

Which factor is important to specify when measuring density?

A) Temperature

Which observation is quantitative?

- A) The brick is red
- B) The gas is odorous
- C) Water freezes at 273 K

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Which unit does not indicate density?

A) g/mm^3

B) kg/L

C) g/cm^3

D) kg/m^2

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A) g/mm^3

B) kg/L

C) g/cm^3

D) kg/m^2

Name the SI unit for
mass

Hint: not the metric base unit

Name the SI unit for
mass

kg

Calculate

Find the volume of a substance with a density of 5 g/cm^3 and a mass of 75 g .

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$$V = 15 \text{ cm}^3$$

A sample of copper has a mass of 84 g and a volume of 12 mL.
What is the density of the sample?

- A) 7 g/mL
- B) 8.0 g/mL
- C) 6.5 g/mL
- D) 7.0 g/mL

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Convert 7.96 L to mL

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7960 mL

Round 0.0074567 to 2 Sig
Figs

Round 0.0074500 to 2 Sig
Figs

0.0074

Round 1.3894×10^{-9} to 3
Sig Figs

Round 1.3894×10^{-9} to 3
Sig Figs

$$1.39 \times 10^{-9}$$

Write 953 000 000 in
scientific notation.

Write 953 000 000 in
scientific notation.

$$9.53 \times 10^8$$

The result of dividing
 10^{-8} by 10^4 is...

The result of dividing
 10^{-8} by 10^4 is...

$$10^{-12}$$

The result of
multiplying 10^8 by
 10^{-4} is...

The result of
multiplying 10^8 by
 10^{-4} is...

$$10^4$$

Which number has
only non-significant
zeros?

A) 0.00078

B) 7.800

C) 0.007800

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B) 7.800

C) 0.007800